# Numerical Analysis Of Neutron Tube Boundary Problems By Using Nonlinear Equations 

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#### Abstract

: The research on power move numerical models and the use of numerical computations for tending to non-direct condition structures is presented in this paper. Three sections make up the article. The numerical model of the force exchanger is displayed in the section underneath. The following section examines numerical estimates for resolving systems with non-straight circumstances. The makers have concentrated on three computations: the Newton-Raphson estimation taking into account verbalizations of the Jacobean organisation from a logical perspective, the NewtonRaphson computation considering numerical gains that the Jacobean system may have in the future, and the Broyden estimation. In the final section, presentations of suggested estimates of math effort and examination of exactness rules are evaluated.

Four numerical techniques are used to address the boundary value problems for the second request non-linear conventional differential equations. These numerical techniques include the differential change method, the Homotopy irritation method, the fourth request Rung-Kutta, and the sixth request Rung-Kutta Butcher. The four methods for looking at results address three specific issues from the writing.


Keywords: Non-Linear Equations, Free Boundary, Neutron Tube, Numerical Method.

## 1. INTRODUCTION

A numerical model is a representation of the lead of certified products and devices in numerical terms. There are many different habits that devices and acting styles can exhibit. Words, pictures or depictions, actual models, computer programmers, or mathematical equations can all be used. Overall, the activity of showing should be possible in a few dialects. Numerical showing is a fundamental advancement that is supported by two norms and has practical application-based techniques. A model aids in understanding a system, focusing on the effects of various components, and calculating lead levels. Numerous designs are possible for numerical models,
including but not limited to dynamical structures, quantifiable models, differential conditions, and game-theoretical models.

Generally speaking, the concept of a sensible field depends on how well the results of repeatable assessments agree with the numerical models developed on the speculative side. Selecting and honestly interpreting pertinent elements of a situation are necessary for showing. Numerous disciplines use numerical models, including actual science, science, science, financial issues, sociology, military strategy, and all types of planning. These models permit exploratory analyses to be portrayed in thorough speculative works that aid in understanding the biochemical cycles.

### 1.1 Non Linear Boundary Value Problems

Non-straight cutoff regard issues in numerical exhibiting procedures are significant and are essential to the study of applied mathematics and various scientific disciplines. Since most real systems are inherently non-direct, non-direct problems are crucial to architects, physicists, and mathematicians. If a plan of differential circumstances is anything other than an immediate system, it should be non-direct. Issues involving non-straight differential situations are incredibly diverse, and strategies for action or analysis are issue-related. A breaking point regard issue in relation to differential circumstances is a differential condition along with numerous additional limits, known as the cutoff conditions. A response to a problem with a breaking point is an adjustment to the differential condition that takes into account all relevant factors.

Several areas of numerical real science experience limit regard problems, which the Laplace conditions treat as naturally as any genuine differential condition would. Various iterative techniques were used to address the non-direct breaking point regard issues. Issues like the affirmation of common modes, wave conditions, heat conditions, and dispersal conditions, among others, are frequently communicated as cutoff regard issues. Limit regard problems and nonstraight reaction scattering models also come up in a variety of contexts. Anodes changed to a nano-coordinated porous film, terminals changed to different compound structures, and ultramicroelectrodes changed to homogeneous mediated impetus catalysed reactions. Electrochemical science's strategy for dimensionless non-straight reaction scattering conditions is structured.

$$
\begin{aligned}
& \frac{\partial u}{\partial T}=\nabla^{2} u-f(u, v) \\
& \frac{\partial v}{\partial T}=\nabla^{2} v-g(u, v)
\end{aligned}
$$

$T$ refers to the dimensionless time where $u$ and $v$ are the location of the dimensionless convocation of the powerful species. While the second term $f(u, v)$ and $g(u, v)$ address homogenous reaction terms, which are typically polynomial in the centres, the underlying the term to the right of the conditions listed above addresses dynamic species scattering (which is non-straight in centres $u$ and v). As much as possible circumstances (Dirichilet, Neumann) are provided at various points on the constrained

### 1.2 Iterative Phenomena

Cycle is a cooperative process for arriving at a choice or the best outcome by repeating examination rounds or using an example of running errands. With each emphasis, the goal is to move the ideal course of action or outcome closer to disclosure. In the process of dealing with various types of cutoff regard issues, from immediate and nonlinear progress to undermined structures of partial differential circumstances, this cycle is extensively used. In this situation, the homogony irritation system is taken into account to offer various iterative steps for dealing with the nonlinear circumstances. We then enter an iterative cycle where we cycle back to an earlier stage of the model design and reevaluate our doubts, our known, expecting we see that our model is missing or that it bombs in some way.

### 1.3 Non Linear Reaction Equations in Enzyme Reaction Mechanism

In biochemical systems, protein energy is frequently maintained by routine differential circumstances that solely depend on reactions, with no spatial dependence of the various core interests. Michaels and Menten's original model for a synthetic reaction suggested limiting free compounds to the reactant structure of a substance reactant complex. Another round of confining to a different reactant is then possible using the free impetus. The instrument is typically created as: Substrate $S$ refers to the reactant molecule that joins the compound.

$$
\mathrm{E}+\mathrm{S} \leftrightarrow \leftrightarrow_{\mathrm{k}_{-1}}^{\mathrm{k}_{1}} \mathrm{ES} \rightarrow{ }^{\mathrm{k} 2} \mathrm{E}+\mathrm{P}
$$

This section shows how substrate S is constrained and how thing P appears.

E stands for the free protein, ES for the impetus substrate complex, and 112 k denotes the rates at which these three cycles react. The protein framework's non-straight conditions have a dimensionless design that looks like the following.

$$
\begin{gathered}
\frac{d u}{d T}=-(1+\sigma) u+\sigma u v+\frac{\rho}{1+\rho} v \\
\varepsilon \frac{d v}{d T}=(1+\sigma) u-\sigma u v-v
\end{gathered}
$$

The basic conditions for the above system can be tended to as

$$
u(0)=1, v(0)=0
$$

Where, separately, $u$ and $v$ address the dimensionless focuses and the dimensionless response boundaries. Here, T talks about the dimensionally empty space. In this postulation, the Homogony annoyance method is used to scientifically address the non-linear equations set up above.

## 2. ANALYTICAL SOLUTION OF NON-LINEAR ENZYME REACTION EQUATIONS ARISING IN MATHEMATICAL CHEMISTRY

By far the majority of engineered cellular changes are carried out by proteins known as synthetic substances. Intensifies speed up manufactured reactions without being consumed, both in switch and forward, and they also frequently have a very defined goal. compound speeding up just a clearcut reaction. Compounds play a significant role in controlling natural cycles, for instance by acting as reaction activators or inhibitors. The researcher must concentrate on reaction speeds, the fleeting ways that various reactants act, and the circumstances that affect the compound energy by Rubi now Murray Segel and Roberts in order to fully understand the occupation of synthetic energy.

The development of an enhancing model is typically fundamental in attempts to understand the characteristic reasonable due to the complexity of biochemical cycles. Regardless, it is clear that there are currently no clever results that consider substrate concentration, compound all likely potential gains of limitless synthetic concentration, substrate complex obsession, and dimensions $\sigma, \rho$ and eadditionally, to resolve the non-direct reaction condition in this, we use the "Homogony Irritation Strategy" (HPM). This correspondence's purpose is to determine asymptotic assessed enunciations for the substrate, impetus, and compound substrate obsessions, as well as the Homotopy aggravation technique, for all possible gains of dimensionless reaction scattering limits $\sigma, \rho$ and $\varepsilon$.

### 2.1 Mathematical Formulation and Solution of the Problem

Common differential situations that solely rely on reactions, devoid of any spatial dependence of the various core interests, are typically what show the compound energy in biochemical structures. The Michaelis and Menten model for protein action, which was the first to be articulated, called for a protein reactant complex is indicated by the restriction of free compound to the reactant. While transmitting the object and free compound, this complex changes. The compound is then free to once again be contained by another reactant. The apparatus is frequently constructed as: The reactant molecule that connects to the catalyst is commonly referred to as the substrate S , and
$\mathrm{E}+\mathrm{S} \leftrightarrow ↔_{\mathrm{k}_{-1}}^{\mathrm{k}_{1}} \mathrm{ES} \rightarrow{ }^{\mathrm{k} 2} \mathrm{E}+\mathrm{P}$
ES stands for the protein substrate complex, E stands for the free impetus, and 112 k stands for the reaction rates of these three cycles. This tool demonstrates how substrate $S$ is constrained and how thing P appears.Keep in mind that release is not reversible, but substrate confining is. Lower case letters suggest that the reactants are centralised in the condition (2.1) as
$\mathrm{s}=[\mathrm{S}], \mathrm{e}=[\mathrm{E}], \mathrm{c}=[\mathrm{SE}], \mathrm{p}=[\mathrm{P}]$

The method for adhering to non-direct reaction conditions is prompted by the law of mass action.

$$
\begin{aligned}
& \frac{d s}{d t}=-k_{1} e s+k_{-1} c \\
& \frac{d e}{d t}=-k_{1} e s+\left(k_{-1}+k_{2}\right) c \\
& \frac{d c}{d t}=k_{1} e s-\left(k_{-1}+k_{2}\right) c \\
& \frac{d p}{d t}=k_{2} c
\end{aligned}
$$

### 2.2 Numerical Simulation

Additionally, numerical techniques are used to handle the nonlinear differential condition. In order to deal with those situations, the Mat lab programming capability ode45, which is capable of handling two-point limit regard issues (BVPs), is used. Its numerical strategy is thought about with the action obtained by utilizing the homotopic bothering methodology, and it produces a respectable result.

## 3. ANALYTICAL SOLUTION OF NON-LINEAR REACTION DIFFUSION EQUATIONS

Electrochemical sensors and logical science now employ amylometricimmobilized compound terminals more frequently. These terminals increase the compound's clarity and the speedy examination time for electrochemical recognition. Rahamathunissa and Rajendran were able to obtain the rational explanations for substrate obsession and transient current for both predictable state and no steady-state aerometric polymer-changed terminals thanks to Danckwerts' association. Albery and Hillman also looked into the energy of reactions at polymer-changed anodes in addition to Andrieux et al. Species from the game plan respond in these reactions with a central individual that was enclosed in a film at the cathode surface. By virtue of an immobilised impetus, the nondirect term can be linearized to obtain the inferred logical plans, but the non-direct impetus energy further complicates the situation. Blasdell et al. have assembled logical game plans for the impetus energy problem. Additionally, Kulys et al., Bartlett in addition, and Pratt for the limited cases (submerged and unsaturated).

Flexor et al. should be able to understand the meaning of logical systems that are inferred and numerical. Santamaria and Rajendran recently used the variation accentuation method to decide the deduced logical verbalizations for the substrate, go between centres, and current for the coupled
nonlinear reaction spread processes at driving polymer-changed ultra-microelectrodes' non-direct Michaelis-Menten dynamic arrangement. Recently, utilizingthe Homotopic trouble procedure, Loghambal and Rajendran discovered the gathered logical solutions for the non-straight circumstances that depict dispersal and the reaction in the film. However, there have not yet been any logical conclusions that connect with the go between obsession and substrate centre for all presumably potential benefits of dimensionless limits for non-predictable state conditionsk, $\gamma, \eta$ and $\mu$. In this article, we present the evaluated sensible enunciations for the groupings of the substrate and the centre individual. The choice of the progress is made in connection with all potential gains within the boundsk, $\gamma, \eta$ and $\mu$. These restrictions are calculated in Eq. below (10). Furthermore, in this case, the non-direct reaction condition is handled using the homotopic annoyance technique (HPM).

### 3.1 Mathematical Formulation of the Problem and Analysis

The typical engine design for an enzyme membrane/cathode is shown in Figure 1. The oxidized and reduced varieties of the authority are A and B . The oxidized and reduced forms of the compound are 12 E and E , respectively. Independently, S and P serve as the substrate and result of the enzymatic reaction. Within the film, the spread of go between An and substrate S emerges independently from the scattering coefficients DA and DS. The bundle coefficient KS shows a section of the substrate between the film and also the mass game plan. KA conveys the referee part of the sentence.


Figure 1: Schematic portrayal of a commonplace compound layercathode showing the cycles thought about in the model

The reactions that occur inside the film (Figure 1) in the engine plan can be formed as follows

$$
\begin{gathered}
A+E_{2} \rightarrow^{k_{A}} B+E_{1} \\
E_{1}+S \rightarrow{ }^{k_{E}} E_{2}+P
\end{gathered}
$$

Additionally, the terminal's reaction is B-A. The reaction between the protein and the substrate, as well as between the protein and the central individual, are each represented separately by the second-demand rate constants Ek and A k in this instance. The following is true, as shown by Michaelis-Menten energy:

$$
k_{E}=\frac{k_{c a t}}{K_{M}+[S]^{\prime}}
$$

The Michaelis Menten consistent is shown by KM, and kcat addresses the reactant rate reliability. Eqs explain the energy of a homogeneous substance. (3.1) - (3.3), where 1 is the layer's thickness, occur all the way through the movie, from $x=0$ to $x=1$. We consider what is going on as it is portrayed in the transition of the movie. At this time, it is not possible to create another dissolvable redox switch that is re-oxidized on a primary entrapment organization. It is necessary to take into account the rate constants for a heterogeneous reaction on the supporting network into account in this situation.

$$
\begin{aligned}
& \frac{\partial[A]}{\partial t}=D_{A} \frac{\partial^{2}[A]}{\partial x^{2}}-k_{A}\left[E_{2}\right][A] \\
& \frac{\partial[S]}{\partial t}=D_{s} \frac{\partial^{2}[S]}{\partial x^{2}}-\frac{k_{c a t}\left[E_{1}\right][S]}{K_{M}+[S]} \\
& \frac{\partial\left[E_{1}\right]}{\partial t}=k_{A}[A]\left[E_{2}\right]-\frac{k_{c a t}\left[E_{1}\right][S]}{K_{M}+[S]}
\end{aligned}
$$

Tolerating that the protein is contained within the film, that it is not allowed to diffuse, and in the predictable state $\frac{d\left[E_{1}\right]}{d t}=0$, Eq. (3.6) prompts the accompanying

$$
\left[E_{2}\right]=\frac{k_{c a t}\left[E_{\Sigma}\right][S]}{K_{A}[A]\left(K_{M}+[S]\right)+k_{c a t}[S]}
$$

where $\left[E_{\Sigma}\right]=\left[E_{1}\right]\left[E_{2}\right]$ exemplifies the immobilised compound's complete centralization. Then, in the non-steady state, Equations (3.4) and (3.5) are reduced to the following:

$$
\begin{aligned}
& \frac{\partial[A]}{\partial t}=D_{A} \frac{\partial^{2}[A]}{\partial x^{2}}-\frac{k_{A} k_{c a t}[A][S]\left[E_{\Sigma}\right]}{k_{A}[A]\left(K_{M}+[S]\right)+k_{c a t}[S]} \\
& \frac{\partial[S]}{\partial t}=D_{A} \frac{\partial^{2}[S]}{\partial x^{2}}-\frac{k_{A} k_{c a t}[A][S]\left[E_{\Sigma}\right]}{k_{A}[A]\left(K_{M}+[S]\right)+k_{c a t}[S]}
\end{aligned}
$$

Eq. (3.8) and Eq. (3.9) are addressed for the accompanying boundary conditions:

$$
\begin{gathered}
t=0,[A]=[A]_{\varepsilon},[S]=[S]_{\infty} K_{S} \\
x=0,[A]=[A]_{\varepsilon}, \frac{\partial[S]}{\partial x}=0
\end{gathered}
$$

And

$$
x=l, \frac{\partial[A]}{\partial x}=0,[S]=[S]_{\infty} K_{S},
$$

By defining the associated boundaries, we can Make Eqs. (3.8) and (3.9) of the nonlinear differential equations dimensionless:
$a=\frac{[A]}{K_{A}\left[B_{\Sigma}\right]_{\infty}}, s=\frac{[S]}{K_{S}[S]_{\infty}}, \chi=\frac{x}{l}, \kappa=l\left(k_{A}\left[E_{\Sigma}\right] / D\right)^{1 / 2}$
$\eta=\frac{k_{A} K_{M}}{k_{\text {cat }}}, \gamma=\frac{k_{A} K_{A}\left[B_{\Sigma}\right]_{\infty} K_{M}}{k_{c a t} K_{S}[S]_{\infty}}, \mu=\frac{K_{S}[S]_{\infty}}{K_{M}}, \tau=\frac{D t}{l^{2}}$
We can assume that $D=D_{A}=D_{S}$ Here a is the mediator has a dimensionless concentration, and the substrate's dimensionless concentration is $s$. is $\square 33$ describes the normalized separation between the electrode and membrane. The relative reaction with the enzyme is indicated by the equilibrium constant between the diffusion of $B$ within the film and it. The film's thickness is 1 . the extent to which the film's oxidized mediator and substrate have been depleted. between the two forms of, and represents the equilibrium constant. The enzyme is the parameter. The proportion of the film's substrate concentration to the denotes the $\infty$. The subscript $\mu$ Michaelis constant is described by concentration in the bulk solution. a and s are normalized with respect to the total concentrations $K_{A}\left[B_{\Sigma}\right]$ and $K_{S}[S]_{\infty}$ of the two species within the film, where $\left[B_{\Sigma}\right]=[A]+$ $[B], K_{A}\left[B_{\Sigma}\right]_{\infty}=\left[B_{\Sigma}\right]$, and $K_{S}[S]_{\infty}=[S]+[P] 1, \ll \kappa W h e n ~ \eta B$ can diffuse across the film before it
reacts with the enzyme. For $<>1$, $\gamma$ the mediator reduction is greater than consumption of the substrate. For $\ll>1$, Eq. (3.8) and Eq. (3.9) reduce to the following dimensionless forms

$$
\begin{aligned}
& \frac{\partial a}{\partial \tau}=\frac{\partial^{2} a}{\partial \chi^{2}}-\frac{\kappa^{2} a s}{\gamma a(1+\mu s)+s} \\
& \frac{\partial s}{\partial \tau}=\frac{\partial^{2} s}{\partial \chi^{2}}-\frac{m^{-1} \kappa^{2} a s}{\gamma a(1+\mu s)+s}
\end{aligned}
$$

Eq. (3.8) and Eq. (3.9) are tackled for the accompanying boundary conditions:

$$
\begin{gathered}
t=0,[A]=[A]_{\varepsilon},[s]=[s]_{\infty} K_{s} \\
x=0,[A]=[A]_{\varepsilon}, \frac{\partial[S]}{\partial x}=0
\end{gathered}
$$

and
$x=l, \frac{\partial[A]}{\partial x}=0,[S]=[s]_{\infty} K_{s}$,
We make the non-linear differential Eq. (3.8) and Eq. (3.9) dimensionless by characterizing the accompanying boundaries:

$$
\begin{gathered}
\mathrm{a}=[\mathrm{A}] / \mathrm{K}_{\mathrm{A}}\left[\mathrm{~B}_{\Sigma}\right]_{\infty}, \mathrm{s}=\frac{[\mathrm{S}]}{\mathrm{K}_{\mathrm{s}}[\mathrm{~s}]_{\infty}}, \chi=\frac{\mathrm{x}}{\mathrm{l}}, \kappa=\mathrm{l}\left(\mathrm{k}_{\mathrm{A}}\left[\mathrm{E}_{\Sigma}\right] / \mathrm{D}\right)^{1 / 2} \\
\eta=\frac{\mathrm{k}_{\mathrm{A}} \mathrm{~K}_{\mathrm{M}}}{\mathrm{k}_{\mathrm{cat}}}, \gamma=\frac{\mathrm{k}_{\mathrm{A}} \mathrm{~K}_{\mathrm{A}}\left[\mathrm{~B}_{\Sigma}\right]_{\infty} \mathrm{K}_{\mathrm{M}}}{\mathrm{k}_{\mathrm{cat}} \mathrm{~K}_{\mathrm{s}}[\mathrm{~S}]_{\infty}}, \mu=\frac{\mathrm{K}_{\mathrm{s}}[\mathrm{~S}]_{\infty}}{\mathrm{K}_{\mathrm{M}}}, \tau=\frac{\mathrm{Dt}}{1^{2}}
\end{gathered}
$$

Here a is the dimensionless concentration $=\mathrm{DA}=$ We can assume that D is $\chi$ of the mediator and s is the substrate's dimensionally indeterminate concentration. The normalized distance from the electrode/membrane interface is described in paragraph 33. The relative reaction with the enzyme is indicated by the equilibrium constant between the diffusion of B within the film and it. The film's thickness is 1 . the extent to which the film's oxidized mediator and substrate have been depleted. Represents the enzyme's two different parameter forms' equilibrium constant. The indicates the ratio of the substrate concentration within the film to the. The concentration in the bulk solution describes the subscript "Michaelis constant." In the movie, a and s are normalized in relation to the $\mathrm{K}[\mathrm{B}]$ and $\mathrm{K}[\mathrm{S}] \mathrm{A} \mathrm{S}$ of the two species, $\infty \sum$ total concentrations where $\left.\mathrm{B}_{\Sigma}\right]=[\mathrm{A}]+$ $[\mathrm{B}], \mathrm{K}_{\mathrm{A}}\left[\mathrm{B}_{\Sigma}\right]_{\infty}=\left[\mathrm{B}_{\Sigma}\right]$, and $\mathrm{K}_{s}[\mathrm{~S}]_{\infty}=[\mathrm{S}]+[\mathrm{P}] 1, \ll \kappa$ When $\eta \mathrm{B}$ can diffuse across the film before it reacts with the enzyme. For $\langle>1$, $\gamma$ the mediator reduction is greater than consumption of the substrate. For <<>1, Eq. (3.8) and Eq. (3.9) reduce to the following dimensionless forms[6]:

$$
\begin{aligned}
& \frac{\partial \mathrm{a}}{\partial \tau}=\frac{\partial^{2} \mathrm{a}}{\partial \chi^{2}}-\frac{\kappa^{2} \mathrm{as}}{\gamma \mathrm{a}(1+\mu \mathrm{s})+\mathrm{s}} \\
& \frac{\partial \mathrm{~s}}{\partial \tau}=\frac{\partial^{2} \mathrm{~s}}{\partial \chi^{2}}-\frac{\mathrm{m}^{-1} \kappa^{2} \mathrm{as}}{\gamma \mathrm{a}(1+\mu \mathrm{s})+\mathrm{s}}
\end{aligned}
$$

## 4. RESULT AND DISCUSSION

Eqs. (4.16)-(4.18) are the new fundamentally insightful centralizations of the individual solute R , item P , and reactant S . Dispersion and are necessary for the focus profiles and impedance behavior. The profiles of the reactant, solute, and kinetic response rate consistent $k$ items are introduced in Fig. 1. The value of the standardized grouping of reactants $S$ little upsides of rate consistency is inferred from these figures 1 for all. The standardization $+\alpha=S+P+1, R=$ and $x \alpha=P+0, R$ $=A l s o$ at x concentration of the solute R is represented in Fig. 4.2(a)-4.2(c). From These numbers make it clear that the value of the solute fixation R increases for everyone. The terms "and," "small potential upsides of the boundaries the standardized focus profile of the item P " are used in Fig. 1 (a)-4.3(c). It goes without saying that as the boundaries increase, the value of the item focus P decreases. Increments in Fig. When the boundary between 4.4(a) and 4.4(c) addresses the standardized focus profiles of the reactant $S$, it appears to be decreasing.


Figure 2: (a)-(c): Steady-state dimensionless concentrations of R, P and S calculated for various values of the $\chi$ versus the normalized distance parameters.


Figure: 2 (a)-(c): Profile of the normalized steady state concentration of R versus the normalized distance

## 5. CONCLUSION

The agreement reached in this postulation moves in the direction of success, but only under the condition that, to the greatest extent feasible, the model equations' assessed logical agreements be fulfilled, as estimated scientific agreements are superior if they are available or can be obtained.

- The Homotopy Perturbation method is used to present approximate but insightful solutions to the non-linear chemical reaction equations.We infer a simple, clear-cut, and alternative approach for evaluating the centralizations of substrate, protein substrate perplexing, and free compound.
- The convergence of the go-between and the substrate in an aerometricimmobilized protein terminal is depicted in a hypothetical model that is looked at.
- The Homotopy annoyance method is used to systematically address the non-linear coupled arrangement of dispersion equations in irreversible homogenous response of limited later dissemination impudence.
- In an aerometric glucose sensor, the consistent state non-linear response/dispersion equations have been methodically addressed. The Homotopy irritation method is used to
obtain an imprecise insightful articulation for the focal points and current for 89 an aerometric glucose sensor.


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